Analysis of the topics covered in Part B of National High School Canadian Chemistry Contest (CCC) Exams since 1995

Please see the exam question papers for the full wording of the questions, which includes details of what students might be expected to cover in their responses. Even where essay titles look similar, they may have required very different approaches.

2021			2015		
	1.	The Aluminum Advantage (mandatory)		1.	Experimental Design: heating up your
	2	New Standards		2	coffee, a question of calorimetry
	۷.	New Standards		2.	,
	3.	Clean Energy?		3.	0 7 1
2020			2014	4.	The potential of fuel cells
	1.	Experimental Design: Determine the Percent Oxalate in an Iron Oxalate Complex Salt (mandatory)	2014	2.	Experimental Design: Producing 25.0 mL of Hydrogen Gas Chemistry is pHun
	2	Van der Waals Forces		3.	What is the Atom?
	۷.	van der waars Poices		4.	8 78
	3.	Chemistry's Role in Creating Solutions to Global Challenges.			Implications for Earth's Energy Resources
2019			2013		
	2.	Experiment Design: heat of combustion of paraffin (mandatory) Chemistry contributions of Gilbert Lewis 2019: The international year of the periodic table		2.3.	Experimental Design: Determine the Enthalpy of Fusion of Ice Why Bother Completing Mendeleev's Periodic Table? Green Chemistry: What is it and why Practice it? The Battery: The Potential of Electrochemistry.
2018			2012		
2010	1.	Experiment Design: Reaction Rates		1.	Experimental Design: K_{ϕ} of sodium chloride
	_	(mandatory)		2.	Three fundamental chemical principles
		Organic Chemistry		3.	Polymer bills
2017	3.	Chemistry		4.	Government restrictions on chemicals available in high schools
4 01/	1.	Experiment Design: Gravimetric Analysis (mandatory)	2011	4	
	2.	Thermodynamics		1.	Experimental Design (Mass and K _a of an unknown weak acid)

- an unknown weak acid) 2. Intermolecular Forces
- 3. Changes to the way atomic masses appear on the periodic table
- 4. Effect of nitrogen dioxide and ozone on air quality

2016

- 1. Experimental Design: identifying an unknown solution
- 2. Catalytic Chemistry
- 3. The future of the periodic table

3. Chemistry and innovation

4. New safety standards in Canada

Analysis of CCC Part B Page 2 of 2 2010 2003 1. My favorite element 1. The Periodic Table 2. VSEPR Theory 2. The Octet Rule 3. Experimental Design (Catalysis) 3. The Kyoto Protocol 4. Ocean Acidification 4. Experiment design (analysis of antacid tablets) 2009 1. Group 14 2002 2. Ethanol 1. Oil (Petroleum) 3. WHMIS 2. The Extraction of Metals 4. Experiment design (preparation of 3. Bonding buffer solutions) 2008 2001 1. Water hardness 1. Air pollution 2. Alternatives to gasoline 2. Catalysts 3. Polymers 3. Isomers 4. Experiment design (determination of an equilibrium constant) 2000 2007 1. The Periodic Table and Periodicity 1. Period 3 2. Water 2. The greenhouse effect 3. Petroleum 3. Chemical equilibrium 4. Experiment design (to determine the 1999 enthalpy change of solution of 1. Polymers 2. Sodium hypochlorite anhydrous copper sulfate in water) 2006 3. Rates of reaction 1. Soaps and detergents 2. Plastics 1998 1. Crude oil 3. Transition elements 4. Experiment design (to determine the 2. Water 3. Forms of elemental carbon enthalpy change of combustion of a fuel called Wizzo) 2005 1997 1. Fire! 1. Extraction of metals and corrosion 2. What is the best kind of water to 2. Biochemicals drink? 3. Electrical conductivity of solutions 3. Gasoline 4. Experiment design (to determine the 1996 rate of reaction between potassium

- 1. Ozone layer
- 2. Alkaline earth metals
- 3. Bonds

1995

- 1. Forms of ionizing radiation
- 2. Acids and bases
- 3. Experiment to determine the enthalpy change of a reaction

1. Acids and Bases

2. Proteins

2004

permanganate and ethanedioic acid)

3. Garbage disposal and recycling

batch of aspirin tablets)

4. Experiment design (to determine the

percentage of acetylsalicylic acid in a